

Co-immobilization of CO₂ Reductase from Potato Tuber Chloroplasts with Thylakoid Membrane for the Production of Formic Acid

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Illumination of isolated chloroplasts from green potato tuber, in absence of HCO_3^- in the medium and in dim 30 lux light, produced a low potential reductant. The generation of this reductant was detected by the characteristic difference spectrum of dark minus light incubated chloroplasts, which showed troughs having maximum at 664, 650, 463 and 392 nm. The spectrum was stable for 1 hr after incubation in dark. But addition of HCO_3^- and incubation at 30 lux light intensity for 30 min abolished the spectrum with concomitant production of formate. Envelope free chloroplasts did not exhibit this difference spectrum under identical conditions. Substitution of envelope free chloroplasts (thylakoid membrane) in the CO₂ reductase assay mixture in place of dithionite showed about 3 fold activation of formate production. Prior treatment of Thylakoid with either DCMU or NADP (10 μM) inhibited formate formation considerably. Thylakoid membrane and CO₂ reductase protein were Co-immobilized in 2% agarose to give a stable system for the biophotolytic production of formate using H₂O and CO₂.

Key Words: Green potato tuber chloroplasts, CO₂ reductase, Thylakoid membrane, Co-immobilization, Bio-photolytic production of formic acid

Introduction

While studies on the production of hydrogen in immobilized systems are in progress, there are some reports on formic acid production by photochemical reduction of CO₂ using synthetic systems. One of such systems where electron transfer from a singlet excited state of aromatic hydrocarbon to another acceptor molecule which in turn reduced CO₂ to formate has been reported by Tazuku and Kitamura (1978). In another system Parkinson and Weaver (1984) achieved the reduction of CO₂ to formate by photogenerated electron in a semiconductor photoelectrode (Indium phosphate) coupled to the enzyme formate dehydrogenase through a mediator methylviologen. Apart from some reports like this, attempts to develop a system producing formate by direct reduction of CO₂ using light energy and water are not many.

Our discovery on the existence of a novel pathway for the direct reduction of CO₂ to formate in green potato tuber chloroplasts, utilizing light energy and water gave a new impetus to this project. Potatoes when exposed to cold temperature (0-4°C) and dim light (30 lux) become green after storage for 8 weeks.

The greening was due to the synthesis of solanine, a toxic alkaloid. We were able to isolate intact chloroplasts from peel tissue which had the complete system for the synthesis of solanidine, the aglycone of the alkaloid (Ramaswamy et al. 1976, Ramaswamy & Nair 1984). The pathway for solanidine formation was established and the primary product of the reaction was identified as formate (Ramaswamy et al. 1983, Nair et al. 1981). Subsequently the CO₂ reductase was isolated and partially purified from these chloroplasts (Arora et al. 1985). These studies proved the way to the development of an immobilized system for the production of formic acid using light energy CO₂ and water which is reported here.

Materials and Methods

Potatoes used for these studies were *C.V. Kufri chandramukhi*. Fresh tubers were brought from local market, cleaned and stored in cold room at 0-4°C under light intensity of 30 lux for 6 to 8 weeks for greening. The green peelings were used for the isolation of chloroplasts.

Chloroplasts were isolated by a modified method of Cockburn et al. (1986) as described by Ramaswamy et al. (1976). All operations were done at 0–4°C and dim light unless otherwise stated.

Activation of CO₂ Reductase

The method of activation of latent CO₂ reductase present in the potato tuber chloroplasts was the same as described in Arora et al. (1985).

Preparation of Envelope-free Chloroplasts

Envelope free chloroplasts i.e. the thylakoid membrane, were prepared by osmotic shock treatment to chloroplasts as described by Ramaswamy and Nair (1984) following Walker's procedure (Walker 1971).

Spectral Evidence for the Formation of a Reductant during Light Irradiation of Chloroplasts

Chloroplasts were suspended in 6 ml of suspension medium (330 mM, sorbitol, 1 mM MgCl₂, 1 mM MnCl₂, 2 mM EDTA, and 50 mM N-2-hydroxyethylpiperazine-N-2-ethane sulfonic acid, Hepes adjusted to pH 7.6 at 20°C with NaOH) to give 200–250 µg chlorophyll/ml. Three ml of this chlorophyll suspension was taken in matched cuvettes of Shimadzu UV 240 spectrophotometer. For taking difference spectrum the absorbance range was adjusted to -0.2 to +0.2 and wave length from 700 nm to 350 nm. The base line was drawn and the chloroplast suspension in the sample compartment was then illuminated for 30 min at 15°C in the presence of 30 lux light under anaerobic condition by passing repurified nitrogen. After incubation the spectrum was recorded against the dark incubated sample. The illuminated sample was then kept in the dark for different time intervals from 1 to 2 hr and then difference spectrum was recorded in between. To the illuminated sample after 1 hr incubation in the dark 7.5 mM NaHCO₃ was added and then incubated for 30 min at 15°C under 30 lux light and anaerobic condition. After the incubation the difference spectrum was recorded. These experiments were repeated with envelope free chloroplasts also.

The chlorophyll content was determined according to the method of Whatley and Arnon (1963).

Reconstitution of CO₂ Reductase to the Thylakoid Membrane

Potato tuber chloroplast CO₂ reductase was activated and then precipitated with acetone. The precipitated proteins were washed with 0.02M Hepes buffer pH 7.5 containing 0.1 M NaCl to remove soluble enzymes and glyoxylate synthetase (Ramaswamy et al. 1983) and suspended in 4 ml of the same buffer. An aliquot of

1 ml was kept for assay of CO₂ reductase. The remaining 3 ml suspension was mixed with washed envelope free chloroplast pellet obtained from 50 g peel tissue to give the reconstituted system. CO₂ reductase activity of this mixture was also determined under same experimental conditions.

Formate formed was estimated by the colorimetric method of Wood and Gest (1957).

Protein concentration was determined by the method of Bradford (1976) using serum albumin as standard protein.

Immobilization of thylakoid and CO₂ Reductase on Agarose to produce Formic Acid Continuously

Agarose was used as the support matrices for the immobilization of CO₂ reductase and thylakoid membrane. Before use, the agarose was sterilized at 15 psi for 15 min. The gels, 2% and 4% were made in 0.02 M Hepes buffer pH 7.5 by heating in a boiling water bath and then cooling to 40°C. The CO₂ reductase preparation and thylakoid membrane preparations from 50 g peel tissue were mixed together, added to the agarose solution and the volume was made up to 10 ml. During addition the solution was stirred well and agarose was allowed to solidify. These gels were packed into two different columns of dimension 9×1 cm after cutting into small pieces with scissor. The packed gel was washed thoroughly with 0.02 M Hepes deoxygenated buffer pH 7.5 containing 5 µg streptomycin/ml till the washings were colourless. All these operations were done at 15°C and 30 lux light intensity.

At this stage, 10 ml 0.02 M Hepes buffer pH 7.5 containing NaHCO₃, 9mM and 50 µg streptomycin was allowed to flow through this column at 12 ml/hr. After collection of 10 ml from the column formate concentration of this eluate was estimated and the column was again washed with NaHCO₃ free buffer. These operations were repeated for number of days.

Results and Discussions

Demonstration of the Formation of Low Potential Reductant by Illumination of Potato Tuber Chloroplasts

The latency of potato tuber CO₂ reductase was established in our earlier studies (Arora et al. 1985). For obtaining an active soluble CO₂ reductase, it was essential to illuminate the chloroplast in the absence of substrate, CO₂, at 30 lux light intensity, which was found to be optimal for the photosynthetic function of these chloroplasts. Absence of the final acceptor of electron, in this case CO₂, may allow accumulation of reducing equivalents and the enzyme CO₂ reductase

may function as a charge accumulator. Formation of a stable low potential reductant in potato tuber chloroplasts upon illumination in the absence of CO_2 was demonstrated by spectral studies. A difference spectrum of dark minus light illuminated chloroplast is shown in figure 1. At 30 min after illumination these chloroplasts displayed a difference spectrum with troughs having maxima at 664, 650, 572, 463, and 392 nm. The light induced spectrum persisted even after 1 hr incubation in the dark. Now the addition of NaHCO_3 and incubation for 30 min, the spectral characteristics disappeared indicating the utilization of accumulated reductant for the reduction of CO_2 . Formate formed could be estimated in this case. The reductant developed during the illumination of the chloroplast in the absence of HCO_3^- was stable even for 2 hr incubation in the dark at 15°C (figure 2). Incubation of chloroplast for such a prolonged period impaired the capacity of HCO_3^- to change the spectrum (figure 2). Similarly removal of CO_2 reductase from chloroplast by preparing envelope free chloroplast did not exhibit characteristic spectrum (figure 3). In this case addition of NaHCO_3 did not affect the spectrum on the other hand an enhancement was observed. These observations are only an indication that some type of reductant accumulates in the chloroplasts altering its

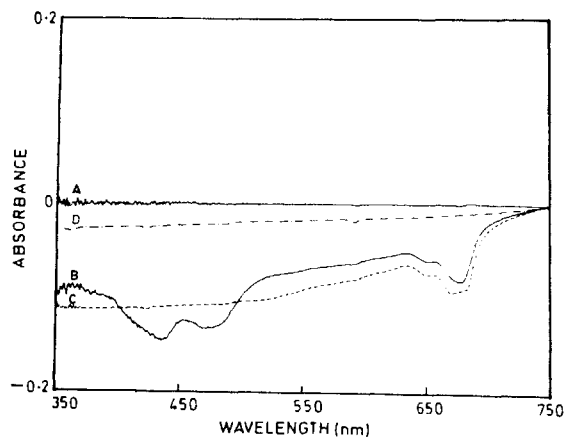


Figure 1 Difference spectra of dark minus light illuminated chloroplasts: Curve A, base lines; curve B after 1/2 hr incubation in light; curve C, after 1 hr incubation in dark; curve D, 1/2 hr incubation after addition of bicarbonate in light

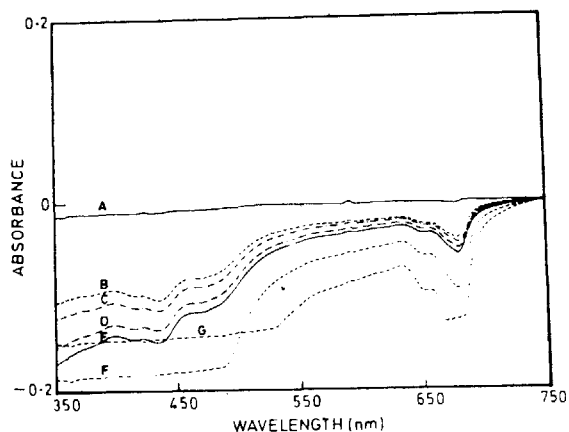


Figure 2 Difference spectra of dark minus light illuminated chloroplast at different time intervals of dark incubation. Curve A Base line; Curve B 1/2 hr in light; Curve C: 1 hr incubation in dark; Curve D, 2 hr incubation in dark; Curve E, bicarbonate added at 0 time after 2 hr dark incubation. Curve F 1/2 hr after adding bicarbonate in light. Curve G 1 hr after adding bicarbonate

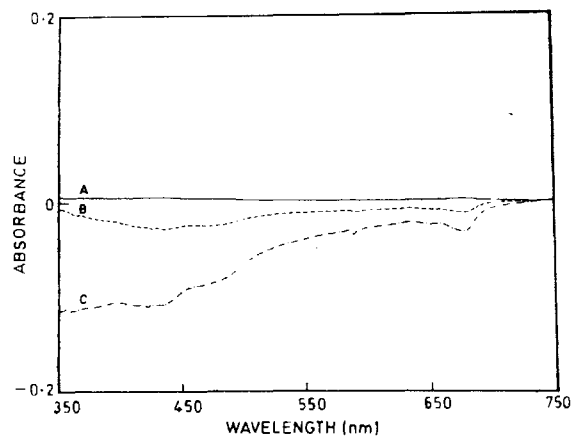


Figure 3 Difference spectra of dark minus light illuminated envelope free chloroplast. Curve A, base line; Curve B, after exposure to light for 1/2 hr; Curve C, after adding bicarbonate

normal absorption spectrum. Significance of these findings in relation to the activation of CO₂ reductase is not well understood at present. Confirmatory proof needs further experimentation and identification of the charge accumulator in the chloroplast. Formation of this type of stable reductant is not observed in any other chloroplasts (Malkin 1976).

Reconstitution of CO₂ Reductase into Isolated Thylakoid Membrane

Isolated CO₂ reductase preparation showed a 3-fold activation by supplementing dithionite in the reaction mixture suggesting that for the optimal functioning of the enzyme there is additional requirement for reducing equivalents (table 1). In the next experiment we have examined whether this requirement for electron donor can be met by supplementing thylakoid membrane free from all soluble enzyme in the reaction system. The thylakoid membrane prepared by osmotic shock did not show any CO₂ reductase activity (table 1). The washed membrane, however, was capable of photosynthetic activity using ferricyanide as electron acceptor (Ramaswamy & Nair 1984). The supplementation of thylakoid preparation in the CO₂ reductase reaction mixture resulted in about 2.7 fold activation of reductase activity (table 1) which showed that CO₂ reductase is reconstituted into thylakoid membrane coupling to electron transfer to CO₂ reduction. Addition of DCMU or NADP strong inhibitors of photolysis of water and photosystem II function in these chloroplasts (Ramaswamy & Nair 1979), inhibited the formate formation. This is suggestive that reducing equivalents derived from water splitting reactions of photosystem II are involved in the reduction of CO₂. The requirement for thylakoid membrane was specific because spinach thylakoids could not replace potato tuber thylakoids.

Table 1 Reconstitution studies on CO₂ reductase with thylakoid membrane

Experiment	Activity (μ moles of formate formed per mg protein)
CO ₂ reductase alone	16.3
CO ₂ reductase plus Na dithionite	48.3
Thylakoid membrane alone	0.0
CO ₂ reductase plus thylakoid membrane	47.5
CO ₂ reductase plus 10 μ M DCMU treated thylakoid	15.0
CO ₂ reductase plus 10 μ M NADP treated thylakoid	0.8

Immobilized System for Formate Production

Since it was possible to reconstitute isolated CO₂ reductase to the washed thylakoid membrane preparation to generate formic acid production, this was thought to be an ideal system for immobilization. The commonly used matrices for immobilizing photosynthetic system are alginate and agarose (Rao & Hall 1984), because they form translucent or transparent gels specially required for light penetration. In this case we have chosen agarose because of its advantage over the other support materials. Here, the thylakoid membrane preparation and CO₂ reductase were Co-immobilized in agarose gel. Two concentration of agarose, 2% and 4% were employed. The efficiency of these two different concentration of gels for the reduction of CO₂ was determined by estimation of formate produced during each operation. The life of the column was determined by continuing the operations till about 40 days. The results of these studies are given in (figure 4). It is evident that 2% gel gave a better yield of formate and its production declined after about 20 days. In both gels maximum activity was obtained after seven days and the maximum yield was retained after for about 2 weeks. Thereafter, the rate of production of formate reduced considerably after 20 days. Continuous illumination of the column at 30 lux light intensity was required for the retention of CO₂ reduction. The design and experimental conditions of a reactor for continuous production of formate using 2% gel for immobilization were described by Arora and Nair (1986).

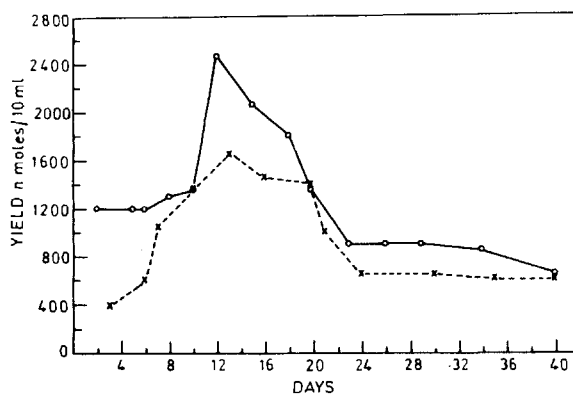


Figure 4 Formic acid production on different days after immobilizing CO₂ reductase and thylakoid preparation in O—O 2% agarose and X—X 4% agarose

The novelty of this system is that it can produce formic acid continuously for about 20 days without any inactivation of the system, if it is washed alternatively after each cycle to remove formic acid completely. Formate trapped in the column may inhibit the binding of CO₂ to the enzyme. The photo-inhibition observed in

the case of higher plant chloroplasts is absent here. This chloroplast system also has an additional advantage as both hydrogen storage and production of organic chemicals from H₂O and CO₂ using light energy.

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